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4.8. The rate of the chemical reaction doubles for an increase of 10 K in absolute temperature from 298 K. Calculate E_a . Ans. 4.9. The activation energy for the reaction, $2HI(g) \rightarrow H_2 + I_2(g)$ is 209.5 kJ mol⁻¹ at 581 K. Calculate the fraction of molecules of reactants having energy equal to or greater than activation energy?

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