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Either  $A = 0$ , or else  $e^{ikL} = 1$ , in which case  $kL = 2n\pi$  ( $n = 0, \pm 1, \pm 2, \dots$ ). But if  $A = 0$ , then  $B e^{-ikL} = B$ , leading to the same conclusion. So for every positive  $n$  there are two solutions:  $\psi_{n+}(x) = A e^{i(2n\pi x/L)}$  and  $u0012L \psi_{n-}(x) = B e^{-i(2n\pi x/L)}$  ( $n = 0$  is ok too, but in that case there is just one solution).

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Introduction to Quantum Mechanics. 2.1. Laws of Quantum Mechanics 2.2. States, observables and eigenvalues 2.2.1. Properties of eigenfunctions 2.2.2. Review of linear Algebra 2.3. Measurement and probability 2.3.1. Wavefunction collapse 2.3.2. Position measurement 2.3.3.

### Lecture notes, Chapter 2. Introduction to Quantum Mechanics

start, in Chapter 3, by examining how many of the central ideas of quantum mechanics are a direct consequence of wave-particle duality—i.e., the concept that waves sometimes act as particles, and particles as waves. We shall then proceed to investigate the rules of quantum mechanics in a more systematic fashion in Chapter 4. Quantum mechanics is

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Quantum mechanics shows that light, along with all other forms of electromagnetic radiation, comes in discrete units, called photons, and predicts its spectral energies (corresponding to pure colors), and the intensities of its light beams. A single photon is a quantum, or smallest observable particle, of the electromagnetic field.

### Introduction to quantum mechanics - Wikipedia

meaningful understanding have not been evaded. It is an introduction to quantum mechanics which. motivates the fundamental postulates of quantum mechanics by considering the weird behaviour of quantum particles. reviews relevant concepts in classical physics before corresponding concepts are developed in quantum mechanics

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