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measurement. a quantitative description  
that includes both a number and a unit.

Scientific notation. an expression of  
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equal to or greater than 1 and less than 10 and  $n$  is an integer. accuracy.

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unit. an expression of numbers in the form  $m \times 10^n$  where  $m$  is equal.... a measure of how close a measurement comes to the actual or tr....

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kathrynphillips4. Chemistry Chapter 3:  
Scientific Measurement. accuracy.  
precision. error. percent error. the  
closeness of a measurement to the true  
value of what is be.... a measure of how  
close a series of measurements are to

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one and.... the difference between the experimental value and the accepted....

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### **Chapter 3 - Scientific Measurement** **- 3.1 Using and ...**

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Using and Expressing Measurements 3.2  
Units of Measurement ... • In chemistry,  
you will often encounter very large or

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very small numbers. ...  $0.00736 = 7.36 \times 10^{-3}$

### 3.1 Using and Expressing Measurements >

### **3.1 Using and Expressing Measurements >**

Section 3: Scientific Notation. The study of chemistry can involve numbers that are very large. It can also involve



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numbers that are very small. Writing out such numbers and using them in their long form is problematic, because we would spend far too much time writing zeroes, and we would probably make a lot of mistakes!

## **Chapter 1: Measurements in Chemistry - Chemistry**

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Scientific Measurement 4 Chapter 3  
Assignment & Problem Set 5. Express  
the following numbers in proper  
scientific notation. a. 0.00567 b. 543000  
c.  $56.7 \times 10^{-4}$  d.  $0.085 \times 10^7$  6. Express  
the following numbers in decimal  
notation a.  $4.94 \times 10^5$  b.  $6.062 \times 10^{-5}$  7.  
Solve each problem and express each  
answer in scientific notation. a.

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### **Scientific Measurement Chapters 3 Assignment & Problem Set**

Chapter 3 - Scientific Measurement - 3.1

Using and Expressing Measurements -

3.1 Lesson Check - Page 72: 13 Answer

To evaluate the accuracy of a measurement, the measured value must be compared to the correct value.

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### **Chapter 3 - Scientific Measurement** **- 3.1 Using and ...**

after reading Lesson 3.1, answer the following questions. Scientific notation 1. Why are numbers used in chemistry often expressed in scientific notation? 2. Circle the letter of each sentence that is true about numbers expressed in

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scientific notation. a. A number expressed in scientific notation is written as the product of a coefficient

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Measurement(continued) Label each of the three following sentences that describes accuracy with an A.Label each sentence that describes precision with a P. \_\_\_\_\_5. Four of five repetitions of a measurement were numerically identical, and the fifth varied from the others in value by less than 1%.

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## **SECTION 3.1 MEASUREMENTS AND THEIR UNCERTAINTY**

About This Chapter The Scientific Measurement chapter of this Prentice Hall Chemistry Companion Course helps students learn the essential lessons associated with scientific measurement. Each of...



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Study Guide for Chapter 3 - Scientific Measurement (Rough outline of the chapter, please use the book, notes & homework to study.) 3.1 Measurements  
Vocab measurement scientific notation  
accuracy precision accepted value

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experimental value error

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