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238 Chemistry: Matter and Change • Chapter 12 Solutions Manual CHAPTER 12 SOLUTIONS MANUAL 7. Challenge Air is a mixture of gases. By percentage, it is roughly 78 percent nitrogen, 21 percent oxygen, and 1 percent argon. (There are trace amounts of many other gases in air.) If the atmospheric pressure is 760 mm Hg, what

States of Matter

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CHAPTER 12 SOLUTIONS MANUAL States

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of Matter Section 12.1 Gases pages 402-410 Practice Problems page 405 1. Calculate the ratio of effusion rates for nitrogen (N₂) and neon (Ne). Rate N / Rate Ne 0.849 Rate N _ Rate Ne $\sqrt{\frac{\text{molar mass Ne}}{\text{molar mass N}}}$ $\sqrt{\frac{20.18}{28.02}}$ 0.849 2.

Chemistry Chapter 12 Solution Manual Stoichiometry

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