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CHAPTER 9 REVIEW Stoichiometry

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SECTION 3 PROBLEMS Write the answer on the line to the left. Show all your work in the space provided.

1. 88% The actual yield of a reaction is 22 g and the theoretical yield is 25 g. Calculate the percentage yield.

2. 6.0 mol of N_2 are mixed with 12.0 mol of H_2 according to the following equation: $N_2(g) + 3H_2(g) \rightarrow 2NH_3(g)$...

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Answers CHAPTER 9 REVIEW
Stoichiometry MIXED REVIEW SHORT
ANSWER Answer the following questions
in the space provided. 1. Given the
following equation: $C_3H_4(g) + xO_2(g)$
 $\rightarrow 3CO_2(g) + 2H_2O(g)$ 4 a. What is the

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value of the coefficient x in this equation? 40.07 g/mol b.

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Chapter 9 Stoichiometry Worksheet Mole review Worksheet. Calculate the molar mass of the following, and then write the molar mass in ratio form. KNO_3 .

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Worksheet Answers

Na_2SO_4 . $\text{Ca}(\text{OH})_2$ $(\text{NH}_4)_2\text{SO}_3$. CuSO_4 .
Moles and Mass. Determine the number
of moles in each of the quantities below.
Use the factor label method. ...

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Chapter 9 - Stoichiometry Chapter 9
focuses on reaction stoichiometry: using
a balanced chemical equation to
calculate the number of grams, moles,
or particles of reactants/products
involved in a...

Chapter 9 Stoichiometry Section 2

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Stoichiometry SECTION 3 PROBLEMS

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Show all your work in the space

provided. 1. 88% The actual yield of a reaction is 22 g and the theoretical yield is 25 g. Calculate the percentage yield.

2.

Chapter 9 Stoichiometry Section 2

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Worksheet

Chapter 9: Standard Review Worksheet

1. Answers will vary. An example is included below: $2\text{H}_2\text{O}_2(\text{aq}) \rightarrow 2\text{H}_2\text{O}(\text{l}) + \text{O}_2(\text{g})$ This describes the decomposition reaction of hydrogen peroxide. Microscopic: Two molecules of hydrogen peroxide (in aqueous solution) decompose to produce two molecules of

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liquid water and one molecule of oxygen gas.

Chapter 9: Standard Review Worksheet

Chapter 9: Standard Review Worksheet
CHAPTER 9 REVIEW Stoichiometry
SECTION 3 PROBLEMS Write the answer
on the line to the left. Show all your work

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in the space provided. 1. 88% The actual yield of a reaction is 22 g and the theoretical yield is 25 g. Calculate the percentage yield. 2.

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Worksheet Answers

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Stoichiometry. SECTION 9.2. PROBLEMS

Write the answer on the line to the left.

Show all your work in the space

provided. 1. The following equation represents a laboratory preparation for oxygen

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Section 2 Answers

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Stoichiometry Section 2 Answers Modern
Chemistry yield of a reaction is 22 g and
the theoretical yield is 25 g Calculate the
percentage yield 2 60 mol of N₂ are
mixed with 120 mol of H [PDF] Chapter 9
Stoichiometry Section 2 Worksheet
CHAPTER 9 REVIEW. Stoichiometry.
SECTION 9.2. PROBLEMS Write the

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