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Problem 2: Property values had plummeted and farms were ruined. Economy was in the dumps. Attempted Solution: The government poured money into the reconstruction of the South but they used taxes to do it, which drained existing resources. Problem 3: African Americans faced challenges with their new freedom like finding land and work and family ...

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Mass % of benzene =  $22/144 \times 100 = 15.28\%$  Mass % of CCl<sub>4</sub> =  $122/144 \times 100 = 84.72\%$  2.2. Calculate the mole fraction of benzene in solution containing 30% by mass in carbon tetrachloride. 1.015 kg of urea solution contains 15g of urea .'. 2.5 kg of solution contains urea =  $15/1.015 \times 2.5 = 37$  g (c ...

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### Answers to Chapters 1,2,3,4,5,6,7,8,9 - End of Chapter ...

CHAPTER . 12 . REVIEW . Solutions . SHORT ANSWER Answer the following questions in the space provided. 1. The following are statements about the dissolving process. Explain each one at the molecular level. a. Increasing the pressure of a solute gas above a liquid solution increases the solubility of the gas in the liquid.

### Solutions - Weebly

Modern Chemistry 1 Solutions CHAPTER 12 REVIEW Solutions Teacher Notes and Answers Chapter 12 SECTION 1 SHORT ANSWER 1. c 2. a 3. b 2. a. alcohol b. water c. the gels 3. The mixture is a colloid. The properties are consistent with those reported in Table 3 on page 404 of the text. The particle size is small, but not too small, and the mixture

### CHAPTER 12 REVIEW Solutions

NCERT Solutions for Class 12 Chemistry Chapter 2 - Solutions This chapter deals with types of solutions, properties, Raoult's law, the concentration of solutions, vapor pressure of liquid solutions, abnormal molar masses, and colligative properties. NCERT Solutions for Class 12 Chemistry Chapter 3 - Electrochemistry

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### **Important Questions for Class 12 Chemistry Solutions Chapter 2**

Chapter 12 Section 2 The Solution Process Factors Affecting the Rate of Dissolution • At higher temperatures, collisions between solvent molecules and solvent are more frequent and of higher energy. This helps to disperse solute molecules among the solvent molecules, and speed up the dissolving process.

### **Chapter 12 Section 2 The Solution Process Factors ...**

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